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TECHNICAL NOTE No: MET.41



### ROYAL AIRCRAFT ESTABLISHMENT

Farnborough, Hants.

### TESTS ON A FLUORIDE FLUX FOR WELDING MAGNESIUM ALLOYS

by

H. BROOKS, B.Sc., A.I.M.

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R.A.E. Technical Note No. Met. 41

July, 1946

### ROYAL AURCRAFT ESTABLISHMENT, FARNBOROUGH

Tests on a fluoride flux for welding magnesium alloys

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H. Erooks, B.Sc., A.I.M. and F/Lt. L. Rakowski, Inz.Chem. (Warsaw)

R.A.E. Ref: Mat. M5/9683/HB/171

### SUMBARY

Tests were made to determine the suitability of a flux for use in welding magnesium-base alloys by the oxyacetylene process. The flux consisted of a mixture of fluorides of lithium, magnesium, calcium and barium. Butt welds were made between pairs of 10G sheet or strip samples to specifications D.T.D.118, R.T.D.120A and D.T.D.259. 'The flux was not satisfactory for welding the magnesium-aluminium-zimo alloys to D.T.D.120A and D.T.D.259 but good welds pould be hade with it in the magnesium-manganese alloy to D.T.D.118. The flux residues caused practically no corrosion when uncleaned joints were exposed to a simulated inland tropical cycle, and joints cleaned by sorubbing and subsequently chromate treated showed no corrosion at all after exposure to these conditions for 28 days.

### Introduction

The sample of flux which has been tested was brought from Germany by a B.I.O.3. investigating town who found that it had been used at I.G. Farbenindustrie Bitterfeld for oxymoetylene welding the magnesium-aluminium-mine alloys, AZa and AZ 855. The flux, known as Elektronfluss IV, was stated to be non-corrosive and non-hygroscopic and to have the following composition:-

Lithium fluoride 22%

Magnesium fluoride 27% Calcium fluoride 18%

Barium fluoride 33%

Walding tests were made with the flux and some of the joints were stored in a humid atmosphere and exemined at intervals to ascertain whether the flux residues had any corrective action.

### Welding tests

These were first made on 10G magnesium-aluminium-zinc alloy sheet

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and extruded strip supplied to specifications D.T.D.120A and 259 respectively and originally it was not intended to carry out tests on the magnesium-manganese alloy to D.T.D.118 since the flux was understood to be unsuitable for this alloy. But welds were made between pairs of 4" × 2" samples, strips out from the parent metal being used as filler material. The flux was mixed with water to form a paste which was brushed on to the filler strip and the top and bottom edges of the joint. Because of the poor action of the flux welding was very difficult on both materials, and although several methods of joint preparation including bevelling were tried, satisfactory penetration could not be obtained. The ineffectiveness of the flux was believed to be partially attributable to its high melting point (stated to be 610°C) so that when welding the magnesium-aluminium-zinc alloys the flux is not completely melted and cannot therefore work efficiently. Since the melting point of the magnesium-manganese alloy is higher than that of the flux it seemed likely that the flux might actually work better on this alloy than on the magnesium-aluminium-zinc alloys which melt at lower temperatures. A few test joints were therefore made in sheet to specification D.T.D.118 and it was found that the flux worked quite well on this material.

### Corrosion tests

Welds in materials to specifications D.T.D.120A and 259 were stored in the laboratory for 6 days and were then exposed to a simulated inland tropical cycle of humidity and temperature for 28 days together with similar welds made with a standard flux A which contains chlorides. Some of the specimens were exposed with the flux residues left untouched on the welds, others were sorubbed in water with a steel brush and then chromated in bath IV of D.T.D.911 before exposure but even this treatment did not completely remove the residues of the fluoride flux. The appearance of typical specimens at the end of these exposure tests is shown in the photographs (Figs.1 and 2) and observations made during the tests are given in Table I. No corrosion tests have been made on welds in D.T.D.118 material since these welds were not made until after the main tests were completed. However it is reasonable to suppose that the flux residues would be no more corrosive to this alloy than to those which have been tested.

### Conclusions

The flux is considered unsuitable for welding magnesium-aluminium-zino alloy sheet and strip in the thicknesses common in aircraft construction but it may be suitable for welding thicker gauges of this type of alloy on which manipulation of the welding rod might be used to assist in breaking up the oxide films. However the flux is suitable for welding thin gauge magnesium-manganese alloy sheet. Where it can be used, the flux possesses the valuable advantage of being practically non-corrosive.

### Attached:

Table I Figs. 1 and 2 (Photographs)

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Hund atmosphere test on fluxes for magnesium alloys

Oxymantylene butt welds, between pieces of 10G sheet or strip, prepared and treated as described below were exposed for 28 days to a 13 day simulated tropical cycle of hundity and temperature: Nirst day: 1409F for 24 hours, relative hundity about 80%, and 15 day simulated tropical cycle of hundity and temperature; Soof and sturies of sours; soof and 95% R.H. for 8 hours; 809F and saturation for 16 hours, with condensation on the

specimens.

ure for the following	28 days	Spots of corrosion all cover; considerable corrosion on underside of weld (this specimen was probably contaminated with chloride flux).	Spots of corrosion all over.	Unchenged	Spots of corresion all over.	Unchanged	Extremely severe corrosion on and near the weld.	Considerable corrosion on the weld.
Observations after exposure for the following periods	ll days	Stots of corrosion all Spots of corrosion all over except on upper corrosion on underside side of weld.  or veld, or weld (this specimen was probably contamina with chloride flux).	Practically no Spots correction on the wold; over, some spots of correction elsewhere.	Unchanged	No corrosion on the S weld; some corrosion selsewhere.	Unchanged	Considerably more corrosion.	Corresion on weld increased.
Appearance after storage in laboratory for 6 days before exposure		Unchanged; welds covered with flux residues.	As Specimen 2.	Unchanged golden-gray film; some hart fused flux residues still left.on welds.	As specimens 2 and 3.	As specimens 12 and 3.	Very considerable corrosion on and near the weld.	A few spots of corrosion on the weld.
Treatment applied after welding		Nema		Soribbed in water with Unchanged steel brush and chromated golden-gray film; in bath IV of D.T.D.911. Some brail flux residues stillar ton welds.	Мотье	Sorubbed in water with As special steel brush and chromated 2 and 3. in bath IV of D.T.D.911.	Иото	steel brush and chromated corrosion on the in bath IV of D.T.D. 911. weld.
Flux used	in welding	Elektron/luss IV						chlorides
Material	Specification	D. T. D. 120A				D.T.D.259	D.T.D.120A	
Specimen	Nos.	N	3	4 and 5	9	7	8	п

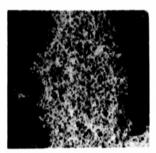


SPEC. 3. EXPOSED "AS WELDED"



SPEC. 4. EXPOSED AFTER SCRUBBING, WASHING AND CHROMATING IN BATH IV OF D.T.D. 911.

WELDED WITH NON-CORROSIVE FLUORIDE FLUX.



SPEC. 8.



SPEC 1

WELDED WITH STANDARD FLUX A CONTAINING CHLORIDES.

APPEARANCE OF UPPER SIDE
OF WELDED D.T.D.120 /10 S.W.G.
SPECIMENS AFTER EXPOSURE TO
HUMID ATMOSPHERE FOR 4 WEEKS

HUMID ATMOSPHERE TEST ON FLUXES FOR MG-ALLOYS

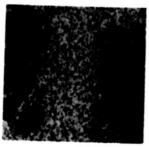


SPEC. 3. EXPOSED "AS WELDED"



SPEC. 4. EXPOSED AFTER SCRUBBING, WASHING AND CHROMATING IN BATH IV OF D.T.D. 911.

WELDED WITH NON-CORROSIVE FLUORIDE FLUX.



SPEC. 8. SPEC.1. WELDED WITH STANDARD FLUX A CONTAINING CHLORIDES.



APPEARANCE OF UNDER SIDE OF WELDED D.T.D.120 / 10 S.W.G. SPECIMENS AFTER EXPOSURE TO HUMID ATMOSPHERE FOR 4 WEEKS

HUMID ATMOSPHERE TEST ON FLUXES FOR MG-ALLOYS

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